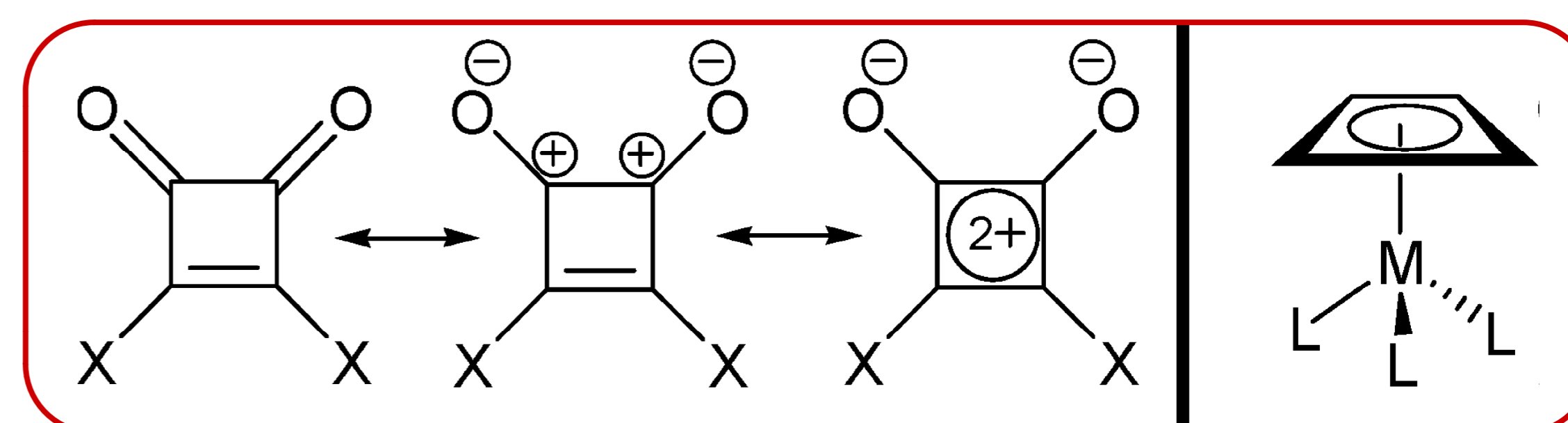


The importance of anion- π interactions in four-membered rings have not been previously described in the literature. Two kinds of rings have been found to be suitable for participation in anion- π interactions:

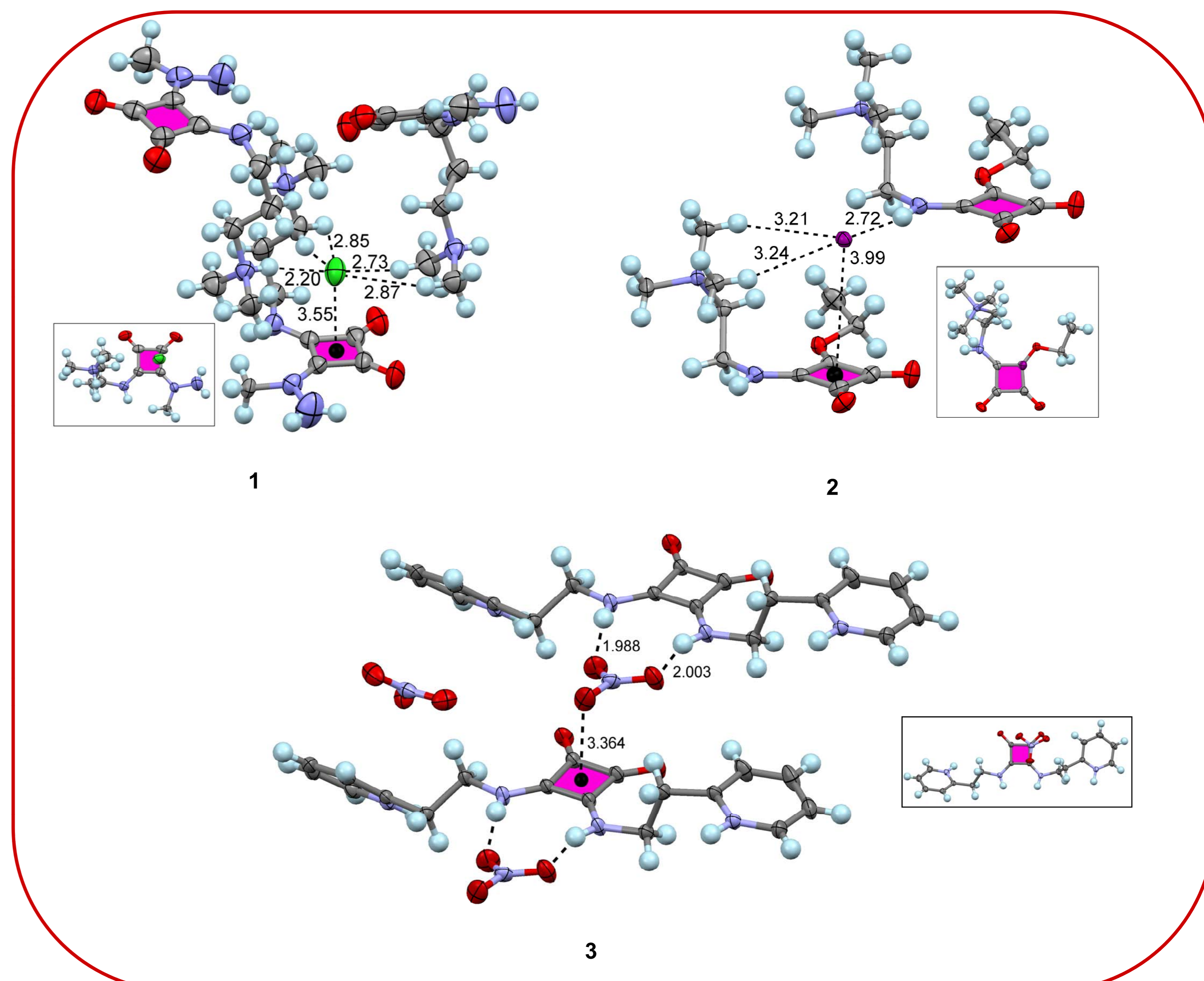
- ✓ different salts of cyclobuten-1,2-dione derivatives
- ✓ η^4 -cyclobutadiene complexes with transition metals.

We have evidenced it by reporting the synthesis and X-ray characterization of three new squaramide salts that exhibit interesting anion- π interactions in the solid state. A search in the CSD has also provided further evidence of its importance. In addition, a high level ab initio investigation on squarate and thiosquarate salts has provided computational support for the experimental observations.

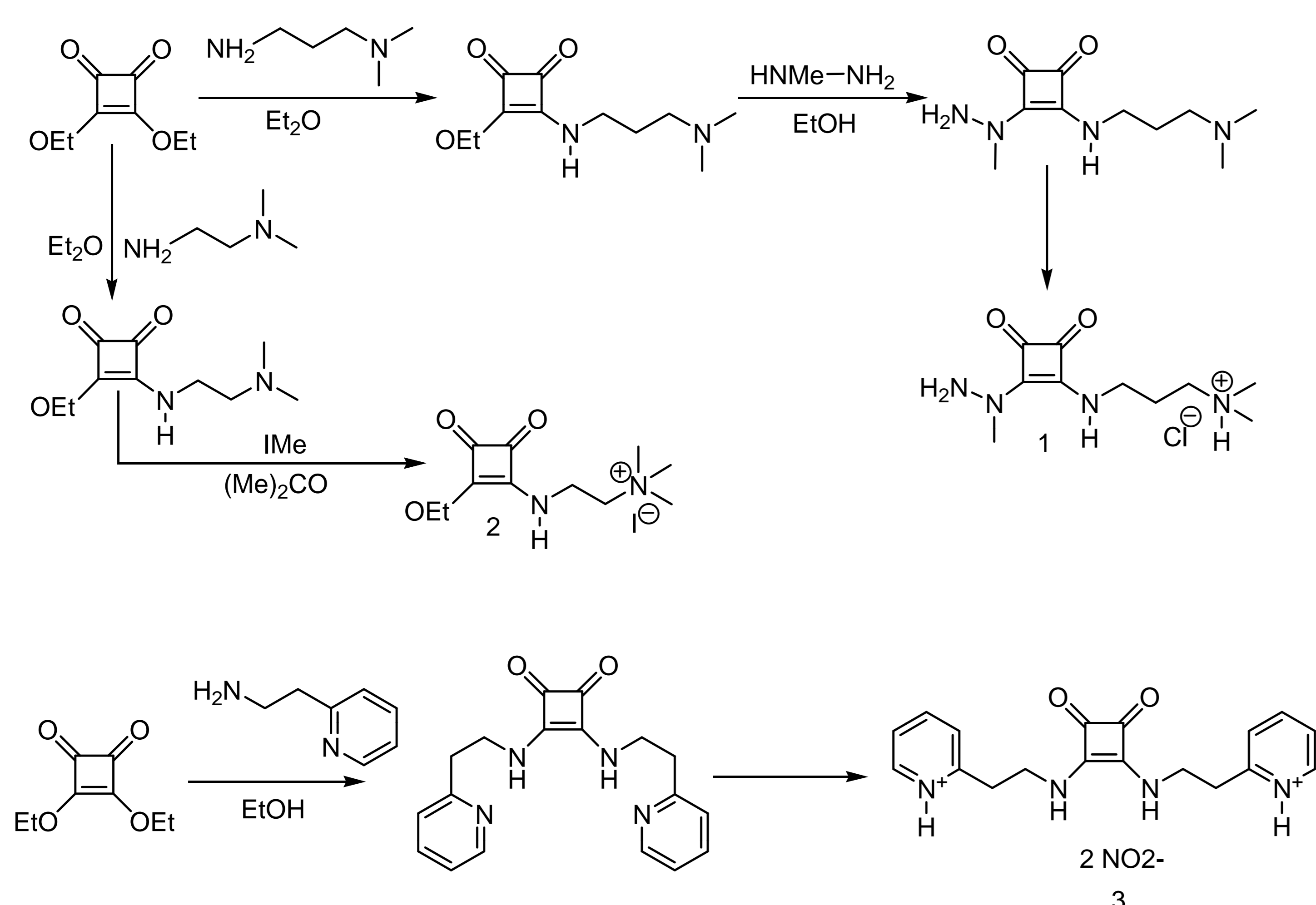
Four-membered rings suitable for anion- π interactions



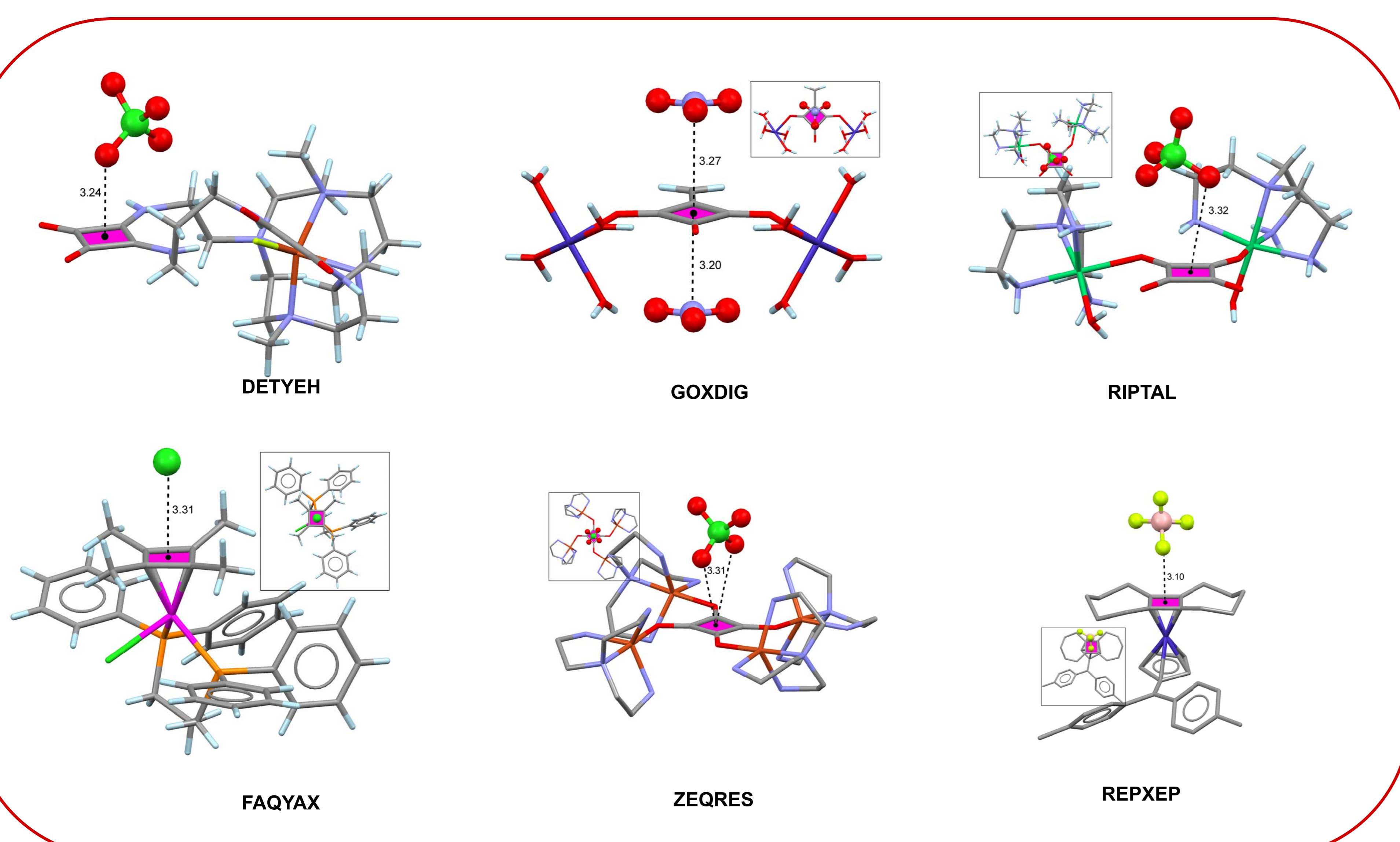
Squaramide X-ray structures featuring anion- π interactions



Squaramides Synthetic Route



Selected structures exhibiting anion- π interactions



Theoretical studies

Schematic representation of model compounds 4–5 and complexes 7–12.

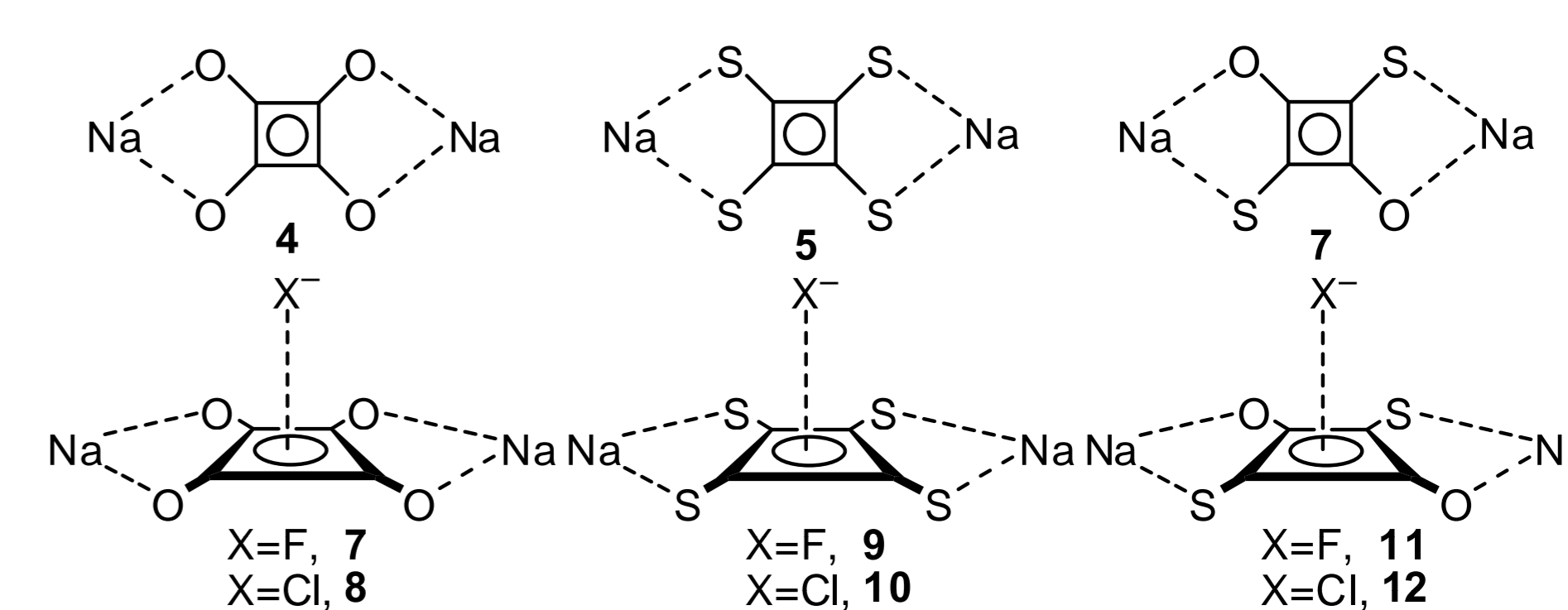


Table. Binding Energies without and with the BSSE Correction (E and E_{BSSE} , kcal/mol, respectively) and Equilibrium Distances (R_e , Å) at RI-MP2/aug-cc-pVTZ Level of Theory for Complexes 7–12 are Summarized.

Complex	E	E_{BSSE}	R_e
7 (4+F ⁻)	-7.27	-4.66	2.596
8 (5+Cl ⁻)	-6.09	-2.99	3.160
9 (5+F ⁻)	-10.00	-5.49	2.319
10 (5+Cl ⁻)	-6.23	-2.68	3.050
11 (6+F ⁻)	-11.17	-7.43	2.425
12 (7+Cl ⁻)	-8.10	-4.13	3.029